



# Sinterability and dielectric properties of $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3\text{--CaTiSiO}_5\text{--Mg}_2\text{TiO}_4$ composite ceramics

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## ABSTRACT

The composite ceramics of  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3\text{--CaTiSiO}_5\text{--Mg}_2\text{TiO}_4$  (BSCT–CTS–MT) were prepared by the conventional solid-state route. The sintering performance, phase structures, morphologies, and dielectric properties of the composite ceramics were investigated. The BSCT–CTS–MT ceramics were sintered at 1100 °C and possessed dense microstructure. The dielectric constant was tailored from 1196 to 141 as the amount of  $\text{Mg}_2\text{TiO}_4$  increased from 0 to 50 wt%. The dielectric constant and dielectric loss of 40 wt%  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3\text{--}10\text{ wt% CaTiSiO}_5\text{--}50\text{ wt% Mg}_2\text{TiO}_4$  was 141 and 0.0020, respectively, and the tunability was 8.64% under a DC electric field of 8.0 kV/cm. The Curie peaks were broadened and depressed after the addition of  $\text{CaTiSiO}_5$ . The optimistic dielectric properties made it a promising candidate for the application of tunable capacitors and phase shifters.

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## 1. Introduction

In recent years,  $(\text{Ba}, \text{Sr})\text{TiO}_3$  (BST) solid solutions have attracted great attention as tunable materials in microwave tunable filed [1]. The dielectric properties of BST ceramics are required to be relatively low dielectric constant (<600), low dielectric loss ( $<6 \times 10^{-3}$ ), less temperature dependence and high electric field tunability [2]. The high sintering temperature (1350 °C) [3] limits the application of BST on multilayer devices. Lots of researchers have fabricated BST–oxide composite ceramics with low dielectric constant, low dielectric loss and flat Curie peaks by mixing BST with non-ferroelectric oxides [4–7]. Among the oxides,  $\text{Mg}_2\text{TiO}_4$  ( $\epsilon_r = 14$ ,  $Q \times f \approx 150,000\text{ GHz}$ ,  $\tau_f \approx -50\text{ ppm/}^\circ\text{C}$ ) is a popular dielectric material applied at microwave frequencies [8,9]. Many researchers have fabricated BST– $\text{Mg}_2\text{TiO}_4$  composite ceramics with good dielectric properties [10,11]. However, the high sintering temperature of approximately 1400 °C needs to be lowered for the application on the LTCC technology [11]. Although nanometer powders prepared by liquid-phase methods [12–15] and sintering agents such as  $\text{B}_2\text{O}_3$  [3],  $\text{Si}_2\text{O}_3$  [16], and  $\text{Li}_2\text{CO}_3$  [17,18] have been used to fabricate low-temperature sintering BST ceramics, reports on the low sintering BST– $\text{Mg}_2\text{TiO}_4$  composite system are rare so far. As was reported [19], the dielectric constant and loss tangent of  $\text{CaTiSiO}_5$

are 45 and  $5 \times 10^{-4}$  measured at room temperature and 1 MHz, respectively. Its low dielectric constant and loss tangent make it available in diluting the ferroelectric response and improving dielectric constant temperature stability of BST ceramics. Moreover,  $\text{CaTiSiO}_5$  is proved to be effective in reducing the sintering temperature of  $\text{Sr}(\text{Ti}_{0.95}\text{Zr}_{0.05})\text{O}_3$  ceramics [20], and doping of  $\text{Ca}^{2+}$  can suppress the dielectric loss and the temperature dependence of BST [21]. Therefore, we choose  $\text{CaTiSiO}_5$  for preparing  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3\text{--CaTiSiO}_5\text{--Mg}_2\text{TiO}_4$  (BSCT–CTS–MT) composite ceramics in order to enhance the sintering performance and dielectric properties of BST– $\text{Mg}_2\text{TiO}_4$  ceramics.

In this paper,  $(90\text{ wt\%--}x)\text{ Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3\text{--}10\text{ wt\% CaTiSiO}_5\text{--}x\text{ Mg}_2\text{TiO}_4$  ( $x = 0, 10, 20, 30, 40, 50\text{ wt\%}$ ) composite ceramics and 50 wt%  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3\text{--}50\text{ wt\% Mg}_2\text{TiO}_4$  for comparison were prepared by solid-state method.  $\text{B}_2\text{O}_3$  and  $\text{Li}_2\text{CO}_3$  were added as sintering aids [18]. The ceramics were sintered densely and then the microstructure and dielectric properties were investigated. The purpose of this work is to enhance the sintering property and the dielectric constant temperature stability of BST materials system, which also expresses low dielectric constant and loss and optimistic dielectric tunability.

## 2. Experimental process

$\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  powders were firstly fabricated by chemical co-precipitation method. Then the powders were calcined in alumina crucible at 750 °C for 4 h in air.  $\text{Mg}_2\text{TiO}_4$  and  $\text{CaTiSiO}_5$  powders were prepared from  $\text{MgO}$ ,  $\text{TiO}_2$  and  $\text{CaCO}_3$ ,  $\text{SiO}_2$ ,  $\text{TiO}_2$  by solid-state method at 1290 °C and 1050 °C, respectively. The

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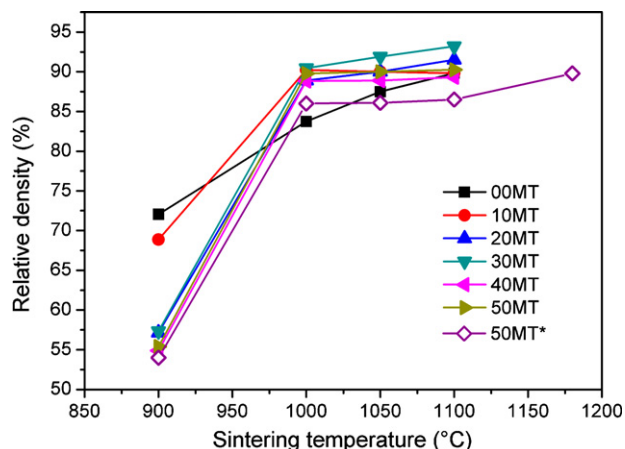


Fig. 1. The relative densities of BSCT-CTS-MT composite ceramics as a function of sintering temperatures.

composites of (90 wt%-x)  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$ –10 wt%  $\text{CaTiSiO}_5$ –x $\text{Mg}_2\text{TiO}_4$  (x = 0, 10, 20, 30, 40, 50 wt%, signed as 00MT, 10MT, 20MT, 30MT, 40MT, and 50MT) and 50 wt%  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$ –50 wt%  $\text{Mg}_2\text{TiO}_4$  (signed as 50MT\*) with an addition of 0.75 wt%  $\text{B}_2\text{O}_3$ –3.75 wt%  $\text{Li}_2\text{CO}_3$  were weighed and then ball-milled for 4 h and dried. The composite powders were added with 8 wt% methylcellulose binder and pressed into pellets. The composite samples were sintered at 900–1180 °C for 2 h.

The phases, morphologies, and elements were analyzed by Rigaku D/Max 2500V/PC X-ray diffractionmeter (XRD) with Cu K $\alpha$  radiation, Hitachi S-4800 scanning electron microscope (SEM), and energy dispersive spectroscopy (EDS). The bulk densities of sintered samples were measured by Archimedes method. The relative densities are calculated by the formula  $\rho_r = \rho_{\text{bulk}} / \rho_{\text{theo}} \times 100\%$ , where  $\rho_r$ ,  $\rho_{\text{bulk}}$ , and  $\rho_{\text{theo}}$  mean the relative density, bulk density, and X-ray theoretical density of the samples, respectively. The dielectric constant and dielectric loss of samples were measured using the Agilent E4980 LCR meter. The temperature dependences of dielectric constant were measured by high–low temperature incubator tank and YY2813 LCR meter at 1 MHz and a temperature range –55 to 125 °C. The tunabilities were tested by HM27004 C-T-V converter and YY2813 LCR meter at 1 MHz and room temperature.

### 3. Results and discussion

#### 3.1. Sintering behaviors, phase structures and morphologies of BSCT-CTS-MT composite ceramics

The relative densities of BSCT-CTS-MT sintered at 900–1180 °C are shown in Fig. 1. It is clearly that the 00MT–50MT reach the relative density of 89.3–93.2% at 1100 °C, while 50MT\* achieves its highest density at 1180 °C. The sintering temperatures are much lower than the pure BST ceramics (1350 °C) and  $\text{BST-Mg}_2\text{TiO}_4$  composite ceramics (1400 °C). The nanometer  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  powders have high specific surface energy so that the ceramics could be sintered densely at a relatively lower temperature. With  $\text{B}_2\text{O}_3$ – $\text{Li}_2\text{CO}_3$  as sintering aids, the liquid-phase sintering mechanism takes effect during the sintering process. Moreover, the sintering temperature of 50MT is lower than 50MT\*, which indicates the addition of  $\text{CaTiSiO}_5$  promotes the sintering densification of BSCT-CTS-MT composite ceramics.

The XRD patterns of 00MT–50MT sintered at 1100 °C and 50MT\* sintered at 1180 °C are shown in Fig. 2. A XRD pattern of pure  $\text{CaTiSiO}_5$  is added in Fig. 2 (named Pure CTS). The  $\text{CaTiSiO}_5$  phase is not detected in all of the patterns. For 00MT, a small amount of  $\text{Ba}_2\text{TiSi}_2\text{O}_8$  is detected besides the perovskite structure  $\text{Ba}_{0.6}\text{Sr}_{0.4}\text{TiO}_3$  phase. The absence of  $\text{CaTiSiO}_5$  phase is possibly because  $\text{CaTiSiO}_5$  reacts with  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  and generates the  $\text{Ba}_2\text{TiSi}_2\text{O}_8$  phase. For 10MT, only the  $\text{Ba}_{0.6}\text{Sr}_{0.4}\text{TiO}_3$  phase is detected, indicating that the  $\text{CaTiSiO}_5$  and  $\text{Mg}_2\text{TiO}_4$  react with  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  and the solid solution is formed, which has little impact on the perovskite structure of  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$ . The radius of  $\text{Mg}^{2+}$  (1.030 Å, CN = 12) is smaller than  $\text{Ba}^{2+}$  (1.610 Å, CN = 12) and  $\text{Sr}^{2+}$  (1.440 Å, CN = 12) [22], so it is easier for  $\text{Mg}^{2+}$  enters into the perovskite structure of  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$ . When the amount of  $\text{Mg}_2\text{TiO}_4$  increases to 20 wt%, the new phase  $\text{BaMg}_6\text{Ti}_6\text{O}_{19}$  appears.  $\text{Mg}_2\text{TiO}_4$  may react with  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  and generate  $\text{BaMg}_6\text{Ti}_6\text{O}_{19}$ .

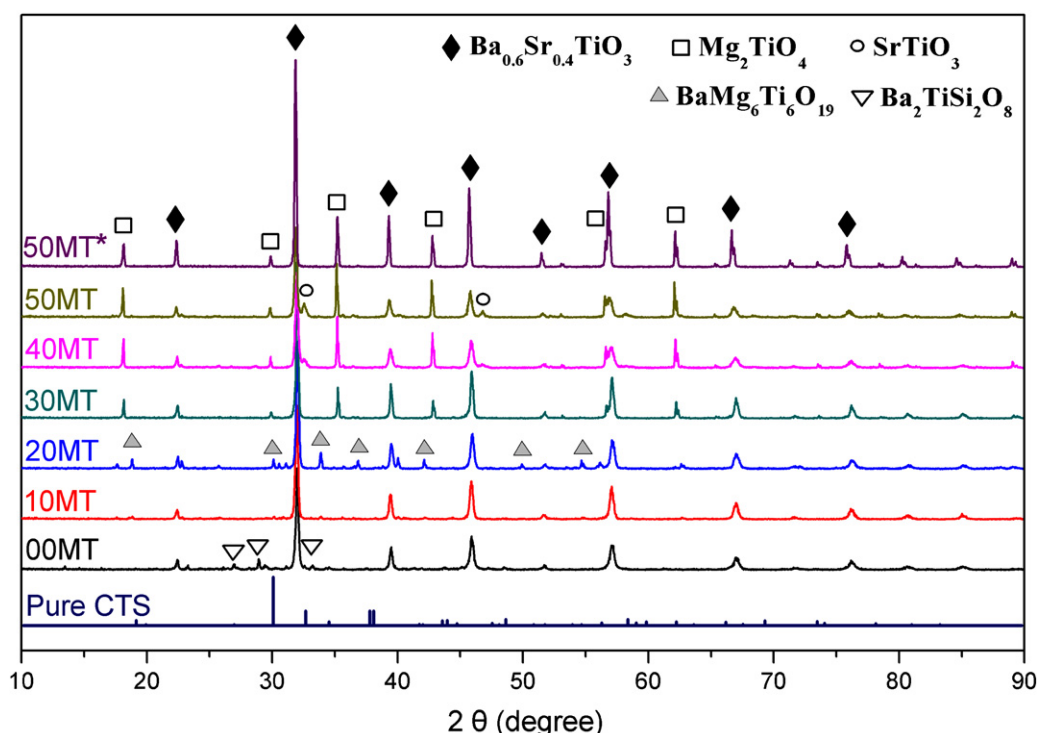
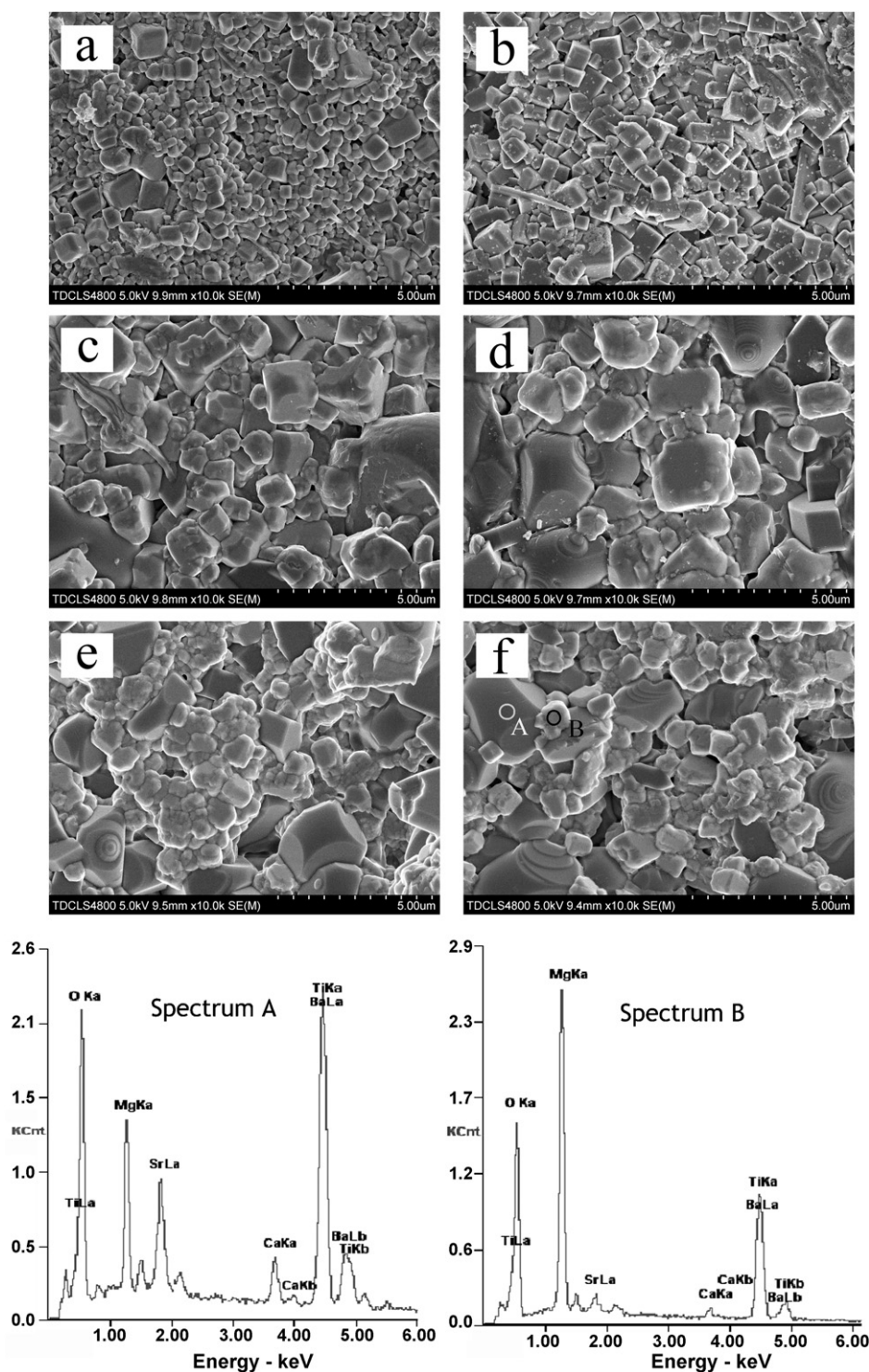


Fig. 2. XRD patterns of 00MT–50MT samples sintered at 1100 °C and 50MT\* sintered at 1180 °C for 2 h.



**Fig. 3.** SEM images of BSCT-CTS-MT composite ceramics and EDS spectra of 50MT ceramics sintered at 1100 °C for 2 h: (a) 00MT; (b) 10MT; (c) 20MT; (d) 30MT; (e) 40MT; (f) 50MT.

Furthermore, more than 30 wt%  $\text{Mg}_2\text{TiO}_4$  is beyond the solution limit ( $\sim 20$  wt%), so there are  $\text{Ba}_{0.6}\text{Sr}_{0.4}\text{TiO}_3$  and  $\text{Mg}_2\text{TiO}_4$  phases and no other phases are observed. Since some  $\text{Ba}^{2+}$  ions remove from  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  lattice, small amount of  $\text{SrTiO}_3$  is detected after 40 or 50 wt%  $\text{Mg}_2\text{TiO}_4$  is added. The XRD pattern of 50MT\* shows coexistence of  $\text{Ba}_{0.6}\text{Sr}_{0.4}\text{TiO}_3$  and  $\text{Mg}_2\text{TiO}_4$  without other phases, revealing that there is no obvious chemical reaction between  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  and  $\text{Mg}_2\text{TiO}_4$ .

The SEM images of BSCT-CTS-MT composite ceramics and EDS spectra of 50MT sintered at 1100 °C are displayed in Fig. 3. It is very clear that two component phases are co-existed and arrangement of grains is compact. EDS analysis in Fig. 3(f) shows that the dark grains with larger size marked as "A" are  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$  and the light small grains marked as "B" are  $\text{Mg}_2\text{TiO}_4$ .  $\text{Mg}^{2+}$  is detected in the spectrum A, and  $\text{Ba}^{2+}$ ,  $\text{Sr}^{2+}$ , and  $\text{Ca}^{2+}$  exist in the  $\text{Mg}_2\text{TiO}_4$  grains, as shown in spectrum B. These ions may enter into the lattice



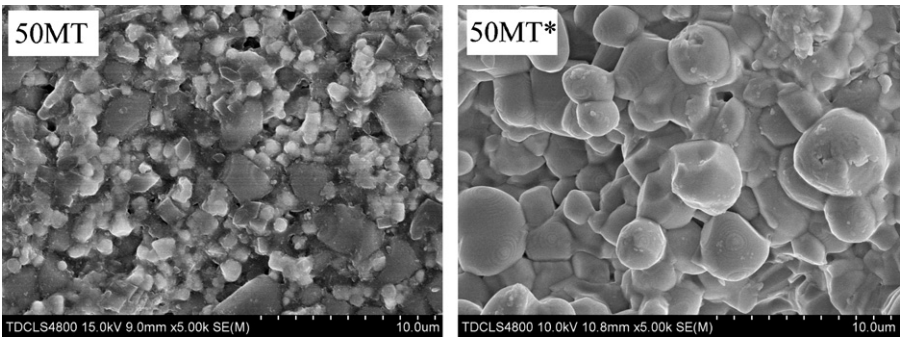


Fig. 4. SEM images of 50MT and 50MT\* sintered at 1100 °C and 1180 °C for 2 h separately.

or exist on the surface of grains. As was reported,  $Mg^{2+}$  can easily substitute  $Ba^{2+}$  or  $Sr^{2+}$  of BST, and  $Mg^{2+}$  ions enter into A site of BST structure initially and then into the B site [9,10]. So it can be inferred that  $Mg^{2+}$  enters into the lattice of  $Ba_{0.55}Sr_{0.4}Ca_{0.05}TiO_3$  grains. It can be proved in the XRD analysis in Fig. 2, in present study, the solubility limit of  $Mg_2TiO_4$  is about 20 wt%. The  $Ba_{0.55}Sr_{0.4}Ca_{0.05}TiO_3$  grain size increases from 0.5 to 2.5  $\mu m$  with the increase of  $Mg_2TiO_4$  content. The SEM images of 50MT and 50MT\* sintered at 1100 °C and 1180 °C are shown in Fig. 4. The morphologies reveal that the  $Mg_2TiO_4$  grains of 50MT grow larger after the addition of  $CaTiSiO_5$  compared to 50MT\*. We can conclude that the  $CaTiSiO_5$  on the grain boundary prevents the growth of  $Mg_2TiO_4$  grain and enhances the densification of BSCT–CTS–MT composite ceramics during the sintering process.

3.2. Dielectric properties of BSCT–CTS–MT composite ceramics

The dielectric properties of BSCT–CTS–MT composite ceramics measured at 20 °C and 2 MHz are summarized in Table 1. The dielectric constant is reduced from 1196 to 141 and the dielectric loss is reduced to the minimum value for 50MT. This reduction is attributed to the addition of dielectric materials  $Mg_2TiO_4$  with low dielectric constant and loss tangent. Furthermore, the electrons are considered to be the cause of leakage loss, which plays an important role in the dielectric loss mechanism. Substitution of  $Ti^{4+}$  by  $Mg^{2+}$  would suppress formation of the  $Ti^{3+}$  so that fewer electrons are created during high temperature sintering, and the leakage loss is reduced as a result. The Curie temperature ( $T_c$ ) which corresponds to the maximum value of dielectric constant shifts to higher temperature as  $Mg_2TiO_4$  content increases. The shift of  $T_c$  results from three reasons. Firstly, below the solid solution limit of  $Mg^{2+}$  in BSCT, the substitutions of  $Mg^{2+}$  in the perovskite leads to change of  $c/a$  ratio and lattice constant of BST ceramics, which affects the  $T_c$  of composite ceramics. Secondly, the grain size of  $Ba_{0.55}Sr_{0.4}Ca_{0.05}TiO_3$  increases from 0.5 to 3  $\mu m$  with increase of  $Mg_2TiO_4$  content. The increasing grain size results in

decrease of internal stress [9], and  $T_c$  shifts to higher temperature. Thirdly, the BST ceramics are solid-state solution of barium titanate and strontium titanate, where a tetragonal ferroelectric phase and a cubic paraelectric phase exist at room temperature. There may be a compositional heterogeneity on microscopic scale in the specimens due to the co-existence of  $Ba^{2+}$  and  $Sr^{2+}$  at the A-site of perovskite structure [23]. The heterogeneity leads to a composition fluctuation in the phase transition temperature, which causes the slightly shift of  $T_c$ . The temperature dependences of dielectric constant for BSCT–CTS–MT composite ceramics measured at 1 MHz are given in Fig. 5. Samples exhibit the diffuse phase transition near Curie temperature and the Curie peaks are depressed and broadened with the containing of paraelectric  $Mg_2TiO_4$  and  $CaTiSiO_5$ . According to the XRD results, the paraelectric phases such as  $Ba_2TiSi_2O_8$ ,  $BaMg_6Ti_6O_{19}$ ,  $Mg_2TiO_4$ , and  $SrTiO_3$  dilute the ferroelectric response of BST ceramics, and more paraelectric phases lead to much flatter Curie peaks.

The tunability is determined by finding the change in dielectric constant at zero-field compared with those dielectric values when there is an applied electric field at room temperature using the formula:

$$\text{Tunability (\%)} = \frac{\varepsilon(E_0) - \varepsilon(E)}{\varepsilon(E_0)} \times 100\%$$
 (1)

where  $\varepsilon(E_0)$  is the zero-field dielectric constant and  $\varepsilon(E)$  is the dielectric constant that results from an applied electric field of  $E$ . The tunabilities of BSCT–CTS–MT composite ceramics sintered at 1100 and 1180 °C at an applied field of 8.0 kV/cm are calculated according to formula (1) and presented in Table 1 and Fig. 6. The figure of merit (FOM) is defined as tunability/ $\tan \delta$ , which is

Table 1  
The dielectric properties of BSCT–CTS–MT composite ceramics measured at room temperature (20 °C) and 2 MHz.

Samples	$\varepsilon_r$	$\tan \delta$	Tunability (8.0 kV/cm) (%)	Figure of merit (FOM)
(at 2 MHz and room temperature)				
00MT	1196	0.0051	10.47	34.90
10MT	1106	0.0024	10.33	40.45
20MT	807	0.0031	8.58	34.32
30MT	1074	0.0027	9.67	35.81
40MT	368	0.0022	8.51	38.68
50MT	141	0.0020	8.64	43.20
50MT*	868	0.0031	19.24	62.06

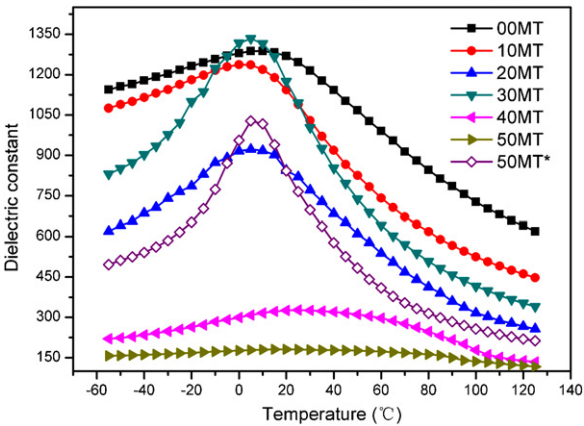


Fig. 5. The temperature dependence of dielectric constant of BSCT–CTS–MT composite ceramics measured at 1 MHz.

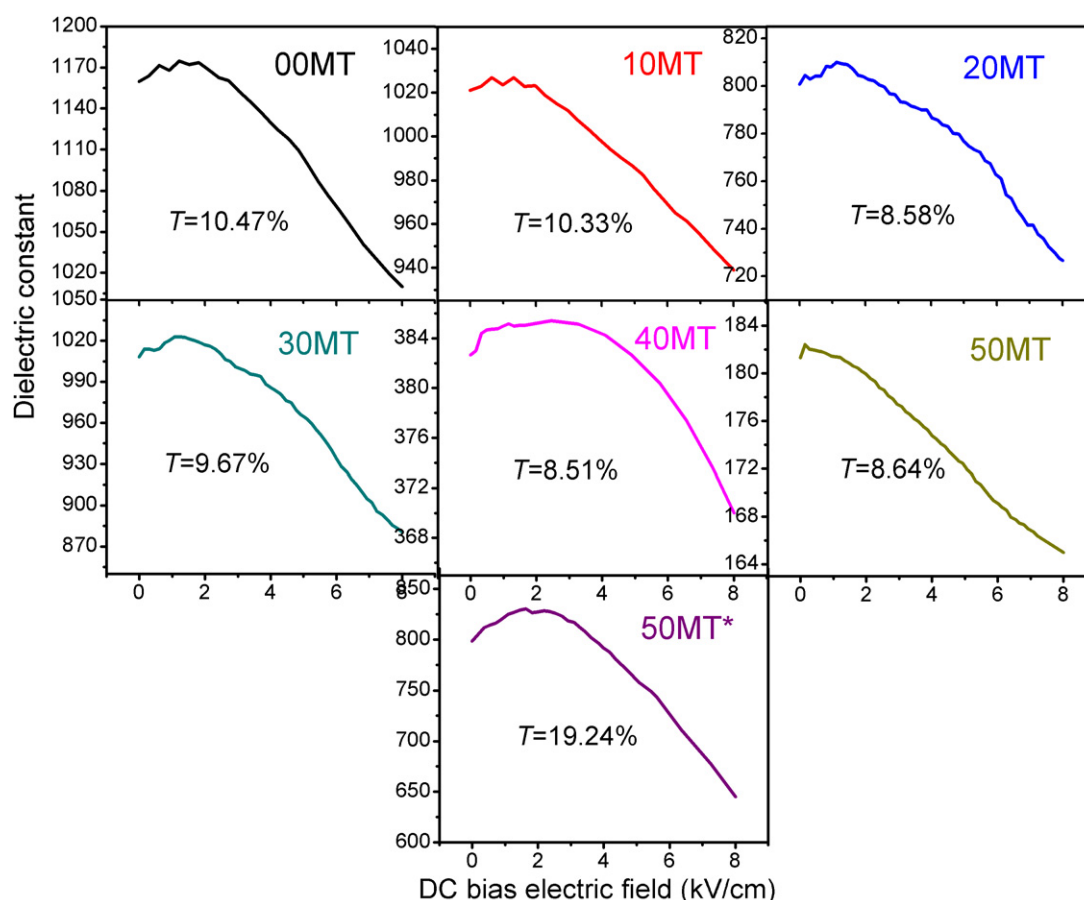


Fig. 6. The DC bias electric field dependence of dielectric constant of 00MT–50MT sintered at 1100 °C and 50MT\* sintered at 1180 °C for 2 h.

usually used to characterize the overall dielectric performance of materials for tunable microwave devices [24]. The DC bias electric field dependence of dielectric constant of BSCT–CTS–MT composite ceramics is shown in Fig. 6. As we all know, the tunability of BST ceramics is associated with the anharmonic interactions of  $\text{Ti}^{4+}$  [25]. The addition of  $\text{Mg}_2\text{TiO}_4$  does not break the connection of Ti–O–Ti bonds, which contributes to the high tunability of BSCT–CTS–MT composite ceramics. The tunability of 50MT\* is much higher than the other samples, which indicates the addition of  $\text{CaTiSiO}_5$  affects the anharmonic interactions of  $\text{Ti}^{4+}$ . The low loss and high tunability result in the high FOM value of 50MT and 50MT\*. This kind of composite ceramics with low loss, high tunability, and excellent temperature stability has great potential for the application of tunable devices.

#### 4. Conclusions

In this work, (90 wt%– $x$ )  $\text{Ba}_{0.55}\text{Sr}_{0.4}\text{Ca}_{0.05}\text{TiO}_3$ –10 wt%  $\text{CaSiTiO}_5$ – $x\text{Mg}_2\text{TiO}_4$  ( $x=0, 10, 20, 30, 40, 50$  wt%) composite ceramics were prepared by solid-state method. The BSCT–CTS–MT composite ceramics were sintered densely at 1100 °C, which was much lower than the sintering temperature of pure BST and BST– $\text{Mg}_2\text{TiO}_4$  composite ceramics. As the  $\text{Mg}_2\text{TiO}_4$  content increased, the dielectric constant and dielectric loss decreased gradually. The Curie temperature shifted to higher temperature and the grain size increased dramatically. The addition of  $\text{CaTiSiO}_5$  was proved to enhance the sintering property and lower the dielectric temperature dependence of composite ceramics. An optimum composite ceramic with dielectric constant 141, loss

tangent 0.0020 at 2 MHz and 20 °C, tunability 8.64% under a DC electric field of 8.0 kV/cm and good temperature stability of dielectric constant was obtained. Both the low dielectric loss and enhanced tunable properties of BSCT–CTS–MT composite ceramics made them useful for microwave tunable applications.

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